Greek Symbols・Atomic Symbols

■Greek Symbols

Uppercase	Lowercase	Pronunciation	Conventional Usage	Uppercase	Lowercase	Pronunciation	Conventional Usage
A	α	alpha	Angle, coefficient	O	o	omicron	
B	β	beta	Angle, coefficient	Π	π	pi	Number π (3.14159…), angle, symbol of volume (uppercase)
Γ	γ	gamma	Angle, weight per unit area, relation (uppercase)				
Δ	δ	delta	Fine difference, density, displacement	P	ρ	rho	Radius, density
E	ε	epsilon	Fine quantity, distortion	Σ	σ	sigma	Stress, standard deviation, summation (uppercase)
Z	ζ	zeta	Variable				
H	η	eta	Variable	T	τ	tau	Time constant, time, torque
Θ	θ	theta	Angle, temperature, time	Y	υ	upsilon	
I	ι	iota		Φ	φ,ϕ	phi	Angle, coefficient, diameter
K	κ	kappa	Rotational radius	X	χ	chi	
Λ	λ	lambda	Wavelength, characteristic value	Ψ	ψ	psi	Angle, coefficient
M	μ	mu	Coefficient of friction 10^{-6} (micron)	Ω	ω	omega	Angular velocity=2πf Ohm=Unit of electric resistivity (uppercase)
N	ν	nu	Oscillation frequency				
Ξ	ξ	xi	Variable				

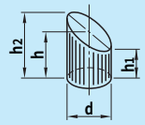
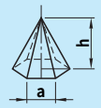
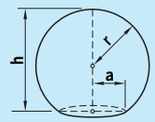
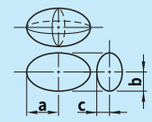
Note: Unless otherwise specified, lowercase letters are the norm.

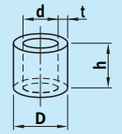
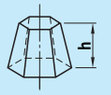
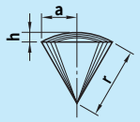
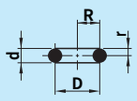
■Atomic Symbols

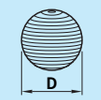
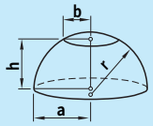
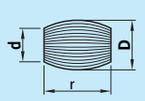
Atomic Number	Name	Symbol	Atomic Number	Name	Symbol	Atomic Number	Name	Symbol
1	Hydrogen	H	36	Krypton	Kr	71	Lutetium	Lu
2	Helium	He	37	Rubidium	Rb	72	Hafnium	Hf
3	Lithium	Li	38	Strontium	Sr	73	Tantalum	Ta
4	Beryllium	Be	39	Yttrium	Y	74	Tungsten	W
5	Boron	B	40	Zirconium	Zr	75	Rhenium	Re
6	Carbon	C	41	Niobium	Nb	76	Osmium	Os
7	Nitrogen	N	42	Molybdenum	Mo	77	Iridium	Ir
8	Oxygen	O	43	Technetium	Tc	78	Platinum	Pt
9	Fluorine	F	44	Ruthenium	R	79	Gold	Au
10	Neon	Ne	45	Rhodium	Rh	80	Mercury	Hg
11	Sodium	Na	46	Palladium	Pd	81	Thallium	Tl
12	Magnesium	Mg	47	Silver	Ag	82	Lead	Pb
13	Aluminum	Al	48	Cadmium	Cd	83	Bismuth	Bi
14	Silicon	Si	49	Indium	In	84	Polonium	Po
15	Phosphorous	P	50	Tin	Sn	85	Astatine	At
16	Sulfur	S	51	Antimony	Sb	86	Radon	Rn
17	Chlorine	Cl	52	Tellurium	T	87	Francium	Fr
18	Argon	Ar	53	Iodine	I	88	Radium	Ra
19	Potassium	K	54	Xenon	Xe	89	Actinium	Ac
20	Calcium	Ca	55	Cesium	Cs	90	Thorium	Th
21	Scandium	Sc	56	Barium	Ba	91	Protactinium	Pa
22	Titanium	Ti	57	Lanthanum	La	92	Uranium	U
23	Vanadium	V	58	Cerium	Ce	93	Neptunium	Np
24	Chromium	Cr	59	Praseodymium	Pr	94	Plutonium	Pu
25	Manganese	Mn	60	Neodymium	Nd	95	Americium	Am
26	Iron	Fe	61	Promethium	Pm	96	Curium	Cm
27	Cobalt	Co	62	Samarium	Sm	97	Berkelium	Bk
28	Nickel	Ni	63	Europium	Eu	98	Californium	Cf
29	Copper	Cu	64	Gadolinium	Gd	99	Einsteinium	Es
30	Zinc	Zn	65	Terbium	Tb	100	Fermium	Fm
31	Gallium	Ga	66	Dysproium	Dy	101	Mendelevium	Md
32	Germanium	Ge	67	Holmium	Ho	102	Nobelium	No
33	Arsenic	As	68	Erbium	Er	103	Lawrencium	Lr
34	Selenium	Se	69	Thulium	Tm			
35	Bromine	Br	70	Ytterbium	Yb			

Note: This table was excerpt from Appendix A (Symbols and Atomic Numbers for Chemical Elements) of ISO 31/8-1980 (Quantities and Units of Physical Chemistry and Molecular Physics) and Appendix C (Names and Symbols for Radioactive Elements) of ISO 31/9-1980 (Quantities and Units of Atomic Physics and Nuclear Physics).

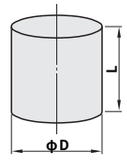
Calculation of Cubic Volume and Physical Characteristics of Metals

Solid	Volume V
Truncated cylinder 	$V=\frac{\pi}{4}d^2h$ $=\frac{\pi}{4}d^2\left(\frac{h_1+h_2}{2}\right)$
Pyramid 	$V=\frac{h}{3}A=\frac{h}{6}arn$ A=Area of base r=Radius of inscribed circle a=Length of a side of a regular polygon n=Number of the sides of a regular polygon
Spherical crown 	$V=\frac{\pi h^2}{3}(3r-h)$ $=\frac{\pi h}{6}(3a^2+h^2)$ a is the radius.
Ellipsoid 	$V=\frac{4}{3}\pi abc$ In case of a spheroid (b=c): $V=\frac{4}{3}\pi ab^2$

Solid	Volume V
Hollow cylinder 	$V=\frac{\pi}{4}h(D^2-d^2)$ $=\pi th(D-t)$ $=\pi th(d+t)$
Truncated pyramid 	$V=\frac{h}{3}(A+a+\sqrt{Aa})$ A,a=Area of both ends
Spherical segment 	$V=\frac{2}{3}\pi r^2h$ $=2.0944r^2h$
Torus 	$V=2\pi^2Rr^2$ $=19.739Rr^2$ $=\frac{\pi^2}{3}Dd^2$ $=2.4674Dd^2$

Solid	Volume V
Circular cone 	$V=\frac{\pi}{3}r^2h$ $=1.0472r^2h$
Sphere 	$V=\frac{4}{3}\pi r^3=4.1888r^3$ $=\frac{\pi}{6}d^3=0.5236d^3$
Spherical belt 	$V=\frac{\pi h}{6}(3a^2+3b^2+h^2)$
Barrel 	When circumferance makes a curve equal to the circular arc, $V=\frac{\pi r}{12}(2D^2+d^2)$ When its periphery makes a curve equal to a parabolic line, $V=0.209r\ (2D^2Dd+1/4d^2)$

■How to Calculate The Weight



Weight W[g] = Volume[cm³] × Specific gravity

Example: Soft steel

ϕD=16 L=50mm, then the weight is:

$W=\frac{\pi}{4}D^2\times L\times$ Specific gravity

$=\frac{\pi}{4}\times1.6^2\times5\times7.85$

≒79[g]

■How to Calculate Dimensional Changes due to Thermal Expansion

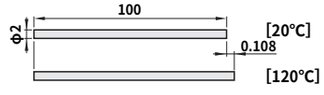
Example: SKD61

When the temperature of a pin whose ϕD=2 and L=100mm raises by 100°C, its dimensional change amount δ is:

δ = Thermal expansion coefficient × Overall length × Temperature change

$=10.8\times10^{-6}\times100mm\times100°C$

=0.108[mm]



■How to Calculate Distortion from Young's Modulus E

Example: The distortion λ obtained when a load of P=1000kgf is applied to a ϕ10×L60 pin (Material : SKD61)

$E=\frac{PL}{A\lambda}$

$\lambda=\frac{PL}{AE}=\frac{1000\times60}{78.5\times21000}$

0.036mm

Crossectional area $A=\frac{\pi}{4}D^2=78.5$

■Physical Characteristics of Metals

Materials	Specific gravity	Young's modulus	Thermal expansion coefficient	
		×10⁻⁶/°C	GPa	{Kgf/mm²}
Soft steel	7.85	11.7	214	21000
NAK80	7.8	12.5	209	20500
SKD61	7.75	10.8	214	21000
SKH51	8.2	10.1	227	22300
Cemented carbide V40	13.9	6.0	551	54000
Cast iron	7.3	9.2~11.8	76~107	7500~10500
SUS440C	7.78	10.2	208	20400
Oxygen free copper C1020	8.9	17.6	119	11700
6/4Brass C2801	8.4	20.8	105	10300
Beryllium steel C1720	8.3	17.1	133	13000
Aluminum A1100	2.7	23.6	70	6900
Duralumin A7075	2.8	23.6	73	7200
Titanium	4.5	8.4	108	10600